This article was downloaded by: [Tomsk State University of Control Systems and Radio]

On: 18 February 2013, At: 11:19

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered

office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/gmcl19

First-Principles Band Structure Calculation for Organic Molecular Crystals

Naoshi Suzuki ^a , Tohru Kawamoto ^a & Masafumi Shirai ^a a Department of Material Physics, Faculty of Engineering Science, Osaka University, Machikaneyama-cho 1-3, Toyonaka, 560, Japan Version of record first published: 05 Dec 2006.

To cite this article: Naoshi Suzuki , Tohru Kawamoto & Masafumi Shirai (1995): First-Principles Band Structure Calculation for Organic Molecular Crystals, Molecular Crystals and Liquid Crystals Science and Technology. Section A. Molecular Crystals and Liquid Crystals, 272:1, 161-165

To link to this article: http://dx.doi.org/10.1080/10587259508055284

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.tandfonline.com/page/terms-and-conditions

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

FIRST-PRINCIPLES BAND STRUCTURE CALCULATION FOR ORGANIC MOLECULAR CRYSTALS

NAOSHI SUZUKI, TOHRU KAWAMOTO and MASAFUMI SHIRAI Department of Material Physics, Faculty of Engineering Science, Osaka University, Machikaneyama-cho 1-3, Toyonaka 560, Japan

Abstract The full-potentail LAPW band calculation in paramagnetic state is carried out for hypothetical square lattices of H₂NO chains by assuming two kinds of stacking of H₂NO molecule: face-to-face stacking and anti-phase alternating stacking. In the former case the ferromagnetic state is never expected to be realized, but in the latter case there is a possibility of appearance of ferromagnetic phase.

INTRODUCTION

In recent decades much interest has been paid to molecular magnetism, in particular to synthesizing organic ferromagnets. As the results, very recently bulk ferromagnetism has been confirmed in several purely organic materils.¹⁻³ But their transition temperatures are quite low ($T_c \sim 1$ K) and hence continued effort has been maid to obtain organic ferromagnets with higher T_c . All the ferromagnetic or antiferromagnetic molecular crystals thus far discovered are well described in terms of Heisenberg Hamiltonian of localized S=1/2 spins. The exchange coupling J has been discussed mainly in the framework of intermolecular charge transfer interaction based on McConnel model⁴, and it has been clarified that the sign of J can be positive (ferromagnetic) if the transfer between SOMO's is smaller compared with that between SOMO and NHOMO (or NLUMO). Low T_c is ascribed to small magnitude of J and it seems not so easy to realize positive and large J. Generally speaking, in 3d magentic systems ferromagnets with quite high T_c are realized in metallic systems. Then we expect that much higher T_c would be obtained by synthesizing metallic organic ferromagnets.

Our purpose of this paper is to perform first-principles band calculations for hypothetical H_2NO crystal as a first step to pursue the possibility of molecular metallic ferromagnets with high T_c . We have chosen the H_2NO crystal because quantum chemical calculation of exchange interaction between two H_2NO molecules has been done by Yamaguchi et al.⁵ According to their results the sign of J depends on stacking of the H_2NO dimer. In case of syn-dimer (face-to-face stacking) J is

negative (antiferromagnetic) while in case of anti-dimer (anti-phase stacking) it is positive (ferromagnetic). Thus it is quite interesting to study how the electronic band structure of H₂NO crystals depends on the stacking of H₂NO molecules.

PROCEDURE OF BAND CALCULATION

For simplicity we have assumed square lattice of H_2NO chains. For the H_2NO chain we consider two kinds of stacking of H_2NO molecules as shown in Figure 1: (A) face-to-face stacking and (B) anti-phase alternating stacking. The former corresponds to the syn-dimer and the latter to the anti-dimer. For the atomic distance within a H_2NO molecule we use 1.04 Å for H-N and 1.23 Å for N-O, and the angle of H-N-H is taken to be 120°.6

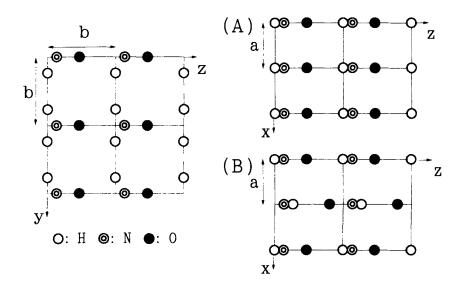


FIGURE 1 Two types of hypothetical structure of H₂NO crystal.

Our band calculations are based on a scalar-relativistic version of the full potential linearized APW (FLAPW) method, i.e. relativistic effects other than the spin-orbit interaction are included. We introduce practically the muffin-tin (MT) sphere around each atom, but the potential is determined fully self-consistently through the whole crystal (both inside and outside the MT spheres). For the exchange-correlation potential we have used the Gunnarsson-Lundqvist type. The core electrons are treated as relaxed. The basis function and the potential inside the MT sphere are expanded up to $\ell_{\text{max}}=7$ and $\ell_{\text{max}}=2$, respectively. The potential outside the MT spheres is expanded in Fourier series of reciprocal lattice vectors G up to $|G|_{\text{max}}=7.8 \text{ Å}^{-1}$. The number of used basis functions are about 360 (720) for the

case (A) (case (B)). We have determined self-consistently the charge density of the crystal using eight k points in the iteration process. The density of states (DOS) has been calculated with a linearly energy-interpolated tetrahedron method.

RESULTS AND DISCUSSION

Actual calculations are performed for the lattice parameters: a=2.75 Å and b=4 Å. Here a denotes the distance between adjacent H₂NO molecules along the chain direction and b is the lattice constant of the square lattice. The electronic DOS's obtained for (A) and (B) cases are shown in Figure 2 (A) and (B), respectively.

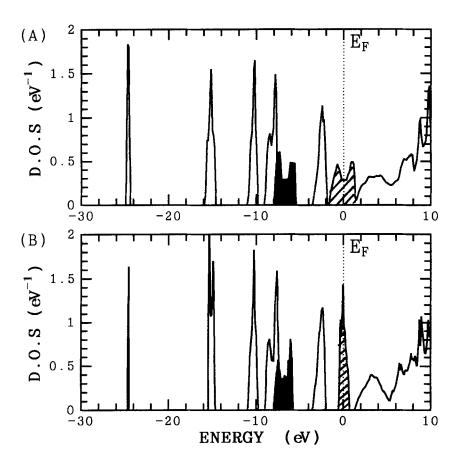


FIGURE 2 The density-of-states (DOS) calculated for the case (A) (face-to-face stacking) and for the case (B) (antiphase alternating stacking). The hatched region denotes the band arising from the anti-bonding states of the π -orbitals of N and O (SOMO) whereas the shaded region that arising from the bonding states.

The electronic band structures of the respective case are characterized as follows:

- (A). Antibonding states of π -orbitals of N and O (SOMO) form a one-dimensional like band and the Fermi level $E_{\rm F}$ lies at the center of this band. Good nesting is realized for wave vector (00π) and therefore a possible magnetic state would be antiferromagnetic, or the system would undergo a CDW (Peierls) transition.
- (B). Compared with the case (A) the electron transfer between SOMO's along the chain direction is much reduced and hence one-dimensionality weakens considerably. Furthermore band degeneracy increases. Thus the density of states at $E_{\rm F}$ increases remarkably and good nesting condition is destroyed. Therefore there is a possibility of appearance of ferromagnetic phase.

The quite different character of the SOMO band in the cases (A) and (B) may be explained as follows. We first note that in each SOMO the amplitude of O π orbital is larger than that of N π orbital. Then the relative spatial configuration between two nearest neihgboring SOMO's can be illustrated for the cases (A) and (B) as shown in Figure 3 (A) and (B), respectively. We easily understand from Figure 3 that the electron transfer between two nearest neighboring SOMO's in case (B) is much smaller than that in case (A). This difference in the electron transfer is regarded as the main origin for the different character of the SOMO bands in the cases (A) and (B).

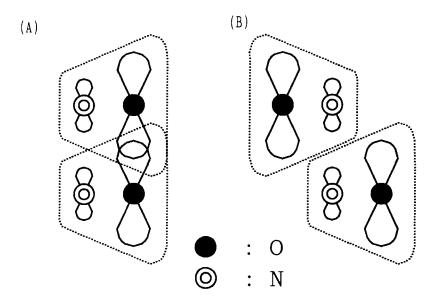


FIGURE 3 The schematic illustration of two SOMO's of the nearest neighboring H₂NO molecules for the case (A) (left) and for the case (B) (right).

By investigating the instability of paramagnetic phase within the HF-RPA approximation we have determined the mangetic phase diagram of the extended Hubbard system in which each atom (molecule) has two different molecular orbitals.⁸ According to our results the ferromagnetic phase may be realized when electron filling is $\frac{1}{4}$ (or $\frac{3}{4}$) and if the transfer between the different orbitals is much larger than that between the same orbitals. The calculated band structure of the case (B) seems to satisfy this condition for ferromagentism. Now we are calculating the spin-polarized (ferromagnetic) band structure for the case (B). The results will be reported elsewhere.

The authors would like to thank Prof. K. Yamaguchi of Osaka University for useful discussion. Thanks are also due to Prof. A. Yanase and Dr. H. Harima of University of Osaka Prefecture for providing us with their FLAPW program. This work is supported by Grant-in-Aid for Scientific Research on Priority Area "Molecular Magnetism" (Area No. 228/06218218) from the Ministry of Education, Science and Culture, Japan.

REFERENCES

- M.Tamura, Y.Nakazawa, D.Shiomi, K.Nozawa, Y.Hosokoshi, M.Ishikawa, M. Takahashi and M.Kinoshita, Chem. Phys. Lett., <u>186</u>, 401 (1991);
 Y.Nakazawa, M.Tamura, N.Shirakawa, D.Shiomi, M.Takahashi, M.Kinoshita and M.Ishikawa, Phys. Rev. B, <u>46</u>, 8904 (1992).
- 2. R.Chiarelli, M.A.Novak, A.Rassat and J.L.Tholence, Nature, 363, 147 (1993).
- 3. T.Nogami, K.Tomioka, T.Ishida, H.Yoshikawa, M.Yasui, F.Iwasaki, H.Iwamura, N.Takeda and M.Ishikawa, Chem. Lett., 29 (1994).
- 4. H.M.McConnell, J. Phys. Chem. Soc., 39, 1910 (1963).
- K.Yamaguchi, M.Okumura, J.Maki, T.Noro, H.Namimoto, M.Nakano, T.Fueno and K.Nakasuji, Chem. Phys. Lett., <u>190</u>, 353 (1992).
- 6. R.Zahradnik and P.Čársky, Theoret. Chim. Acta (Berl.), 27, 121 (1972).
- 7. O.Gunnarsson and B.I.Lundqvist, Phys.Rev.B, 13, 4274 (1976).
- 8. T.Kawamoto, M.Shirai and N.Suzuki, Synthetic Metals, to be published.